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Stereoselective Synthesis of β -L-Rhamnopyranosides

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In spite of remarkable progress in O-glycoside bond forming reactions, the stereoselective synthesis of β -L-rhamnopyranosides $(\beta$ -L-Rha) remains challenging. Although its difficulty derives from structural features similar to β -D-mannopyranosides (β -D-Man) (i.e., equatorial, 1,2-cis),¹ the formation of β -L-Rha is by far more difficult. The direct glycosylation strategy employed for β -D-Man is not effective for β -L-Rha because the 6-deoxy structure of Rha excludes the use of a donor with 4,6-O-cyclic (e.g., benzylidene) protection,² which is essential for the stereoselective β -D-Man formation. Although β -D-Rha may be formed from β -D-Man through deoxygenation,³ this strategy is not suitable for β -L-Rha because of the limited availability of L-Man. In spite of various attempts using structurally modified donors, such as 2,3-O-carbonate,⁴ 2,3-*O*-alkylidene,⁵ 3,4-*O*-carbonate,⁶ 2-ulosyl⁷ or 2-*O*-sulfonate,⁸ 1,2-O-stanylene acetal,9 and allyl-mediated IAD,10 there was limited selectivity. We report herein the unprecedented stereoselective synthesis of β -L-Rha through intramolecular aglycon delivery (IAD).

Our strategy exploits 2-naphthylmethyl (NAP) ether^{11,12} to make a temporary linkage between donor and acceptor as a mixed acetal. We recently found that NAP ether is highly favorable as a tether for IAD. NAP-mediated IAD was found to be versatile, providing β -D-mannopyranosides, β -D-arabinofuranosides, and α -D-glucopyranosides in high yields and complete selectivity.¹³ β -L-Rha has been discovered in various bacterial polysaccharides, such as Sphaerotilus natans,14 Azotobacter beijerinckii TNM1,15 Salmonella serogroup,16 Shigella boydii type 18,17 and Vibrio cholerae NRT36S.¹⁸ These polysaccharides have $(1\rightarrow 3)$ - β -L-Rhap substructures that are linked to Glc⁰⁻⁴, Glc⁰⁻³, Man⁰⁻², Rha⁰⁻⁴, and GlcNAc⁰⁻⁴, respectively. From examination of these structures, the corresponding derivatives 9, 16, 17, 18, and 19 were chosen as acceptors for this study (Table 1). As rhamnosyl donors, 2-O-NAP equipped thioglycosides 4, 5, and 8 were designed, which were all synthesized from thiorhamnoside 1.

Thus, compounds **4** and **5** were prepared through 3,4-*O*-diacetal **2** and 2-*O*-NAP ether **3**, while preparation of compound **8** was conducted via monobenzyl ether **6**, which was regioselectively alkylated to give compound **7** (Scheme 1). The capacity of these donors was compared using the glucose derivative **9** as an acceptor. The formation of mixed acetals (MAs) with 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) under anhydrous conditions afforded **10** (89%, diastereomeric ratio 12:1), **11** (75%, diastereomeric ratio 7:1), and **12** (93%, diastereomeric ratio 25:1). To our delight, subsequent IAD under standard conditions with MeOTf and 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) proceeded smoothly. From **10** and **11**, β -L-rhamnosides **13** (72%, $J_{C1-H} = 162 \text{ Hz})^{20}$ and **14** (56%, $J_{C1-H} = 164 \text{ Hz})^{20}$ were obtained as single isomers, after acidic treatment and acetylation (Protocol A).

A more favorable result was obtained with 3-O-TMS protected MA **12** (Protocol B). In this case, the TMS ether intramolecularly trapped the transiently generated benzylic cation to give the



 a Conditions: (a) DDQ, MS 4Å, CH₂Cl₂, 0 °C, 4 h; (b) MeOTf, DTBMP, MS 4Å (CH₂Cl)₂, 48 h. b Determined by ¹H NMR after isolation.

Scheme 1. Synthesis of Rhamnosyl Donors 4, 5, and 8^a



^{*a*} Conditions: (a) (i) 2,3-butanedione, CH(OMe)₃, CSA, MeOH, 86%, (ii) NAPBr, NaH, DMF, 96%; (b) TFA, H₂O, CH₂Cl₂, 97%; (c) for **4**, BnBr, NaH, DMF, 88%; (d) for **5**, BzCl, DMAP, CH₂Cl₂, Et₃N, 92%; (e) see ref 19; (f) NAPBr, NaH, DMF, 71%; (g) TMSCl, imidazole, DMF, 93%.

naphthylidene cyclic acetal **15**,²¹ which was obtained in 83% yield $(J_{C1-H} = 164 \text{ Hz})^{20}$ (Scheme 2).

The operationally simple IAD of crude **12** was attempted (Protocol C, Table 1), which gave **15** in a satisfactory yield of 68% based on **9**. The stereochemistry of the acetal carbon was assigned as endo, based on a ¹H NMR NOE experiment. Namely, NOE correlations were observed between the naphthylidene acetal proton



^a Conditions: (a) DDQ, MS 4Å, CH₂Cl₂; (b) Protocol A for MA 10 and 11, (i) MeOTf, DTBMP, MS 4Å, (CH₂Cl)₂; (ii) TFA, CH₂Cl₂, then Ac₂O, DMAP, pyridine; Protocol B for MA 12, only (i). ^b Determined by ¹H NMR after isolation.

Scheme 3. Determination of Stereochemistry for 15 as endo



Scheme 4. Complete Stereoselective Synthesis of a Trisaccharide



and the ring protons at C-2 and C-3 in the rhamnopyranoside (Scheme 3).²

Having achieved the stereoselective synthesis of β -L-Rhap(1 \rightarrow 4)Glc (15), acceptors that correspond to Glc^{3-OH} (16), Man^{2-OH} (17), Rha^{4-OH} (18), and GlcNAc^{4-OH} (19) were examined. As expected, reactions with 8 gave desired products 20 [β -L-Rhap $(1 \rightarrow 3)$ Glc], 21 $[\beta$ -L-Rhap $(1\rightarrow 2)$ Man], 22 $[\beta-L-$ Rhap(1 \rightarrow 4)Rha], and **23** [β -L-Rhap(1 \rightarrow 4)GlcN] in 64–71% yields (Table 1).

With the aim of synthesizing bacterial glycans, regioselective opening of the naphthylidene acetal was conducted to liberate the 3-OH of β -L-Rhap. DIBAL-H was suitable for this purpose, solely giving 2-O-NAP ether 24 from 15. Subsequent α -L-rhamnopyranosylation of 24 with trichloroacetimidate 25²³ cleanly afforded the desired trisaccharide 26 in high yield, which was deprotected under standard conditions to achieve the synthesis of 27, corresponding to the substructure of S. natans polysaccharide¹⁴ (Scheme 4).

In conclusion, application of the novel naphthylmethyl ethermediated IAD toward the stereoselective construction of β -L-Rhap was achieved with various acceptors in good yield. The complete stereoselective synthesis of a trisaccharide, α -L-Rhap-(1 \rightarrow 3)- β -L-Rhap- $(1 \rightarrow 4)$ Glcp from S. natans, was successfully accomplished, clearly suggesting the utility of NAP-IAD for β -L-rhamnopyranosylation.

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Supporting Information Available: Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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